



ANSWER SHEET Sample Test v.2

Subject: Chemistry

Date:

City:

Part A: Multiple Choice Questions

1	a	b	c	d
2	a	b	c	d
3	a	b	c	d
4	a	b	c	d
5	a	b	c	d
6	a	b	c	d
7	a	b	c	d
8	a	b	c	d
9	a	b	c	d
10	a	b	c	d
11	a	b	c	d
12	a	b	c	d
13	a	b	c	d
14	a	b	c	d
15	a	b	c	d
16	a	b	c	d
17	a	b	c	d
18	a	b	c	d
19	a	b	c	d
20	a	b	c	d

21	a	b	c	d
22	a	b	c	d
23	a	b	c	d
24	a	b	c	d
25	a	b	c	d
26	a	b	c	d
27	a	b	c	d
28	a	b	c	d
29	a	b	c	d
30	a	b	c	d
31	a	b	c	d
32	a	b	c	d
33	a	b	c	d
34	a	b	c	d
35	a	b	c	d
36	a	b	c	d
37	a	b	c	d
38	a	b	c	d
39	a	b	c	d
40	a	b	c	d

Part B: Short Answer Questions

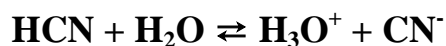
1. Indicate the number of electrons, protons and neutrons in the isotope ${}_{35}^{79}\text{Br}$.

No. of electrons = 35

No. of protons = 35

No. of neutrons = 44

2. In the following equation, identify the acid, base, conjugate acid and conjugate base:



acid HCN

conjugate base CN^-

base H_2O

conjugate acid H_3O^+

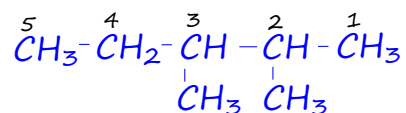
Identify the conjugate acid-base pairs:

HCN / CN^-

and

$\text{H}_2\text{O} / \text{H}_3\text{O}^+$

3. Write the structural formula of the compound 2,3-dimethylpentane.



4. Show the equation and name the product formed when propene reacts with HBr.

