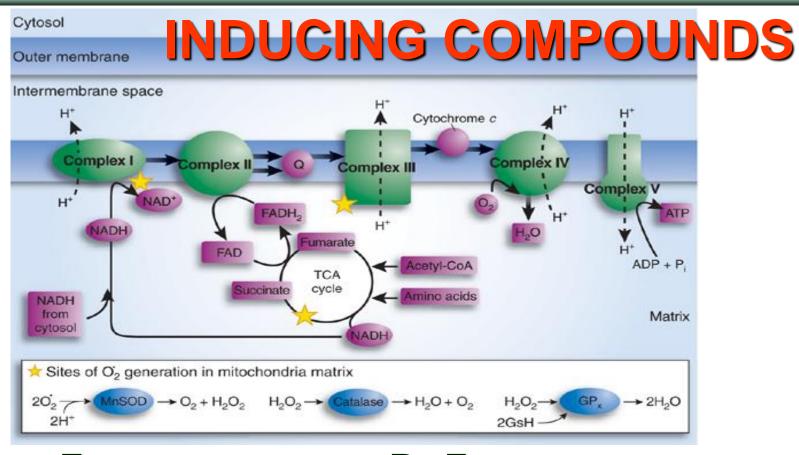


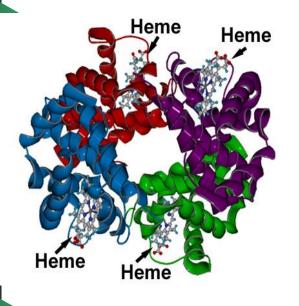
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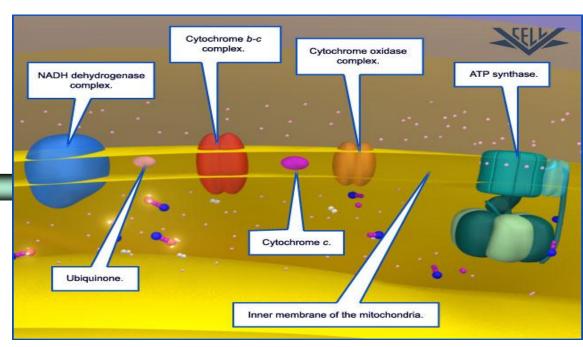
CENTER FOR DISTANCE LEARNING

CELLULAR ASPHYXIA –



Лектор: доц. д-р В. Данчева, дм





 This group includes: carbon monoxide, cyanide, hydrogen sulfide, etc. They induce hypoxia or anoxia by binding of the hemoglobin (carboxyhemoglobin, CO) or by binding of the ferric ion atom in cytochromes (cyanides)

Mitochondrial Toxins

Carbon monoxide

Cyanide

Hydrogen sulfide



 Natural production of carbon monoxide is estimated to be about 10 times the amount produced by manmade sources. Oxidation of methane is the highest source of CO in the atmosphere and may produce 3 billion metric tons of CO annually in the Northern Hemisphere alone. Large amounts of CO are released from the ocean each year by the float cells of kelp.

organic materials Anaerob conditions methane
Biological decay

Natural sources of carbon monoxide



Indirect sources:
mud, bogs
anaerob
conditions
methane
formation from
the decay of
organic materials

The surface of oceans is supersaturated in carbon monoxide

Decay of chlorophyll in the soil

Carbon monoxide poisoning is now the most common of all poisonings in industry and is responsible for more than half of the poisoning fatalities reported each year. It is leading agent of lethal inhalation in the United States, responsible for 6000 accidental and suicidal deaths per year. ~40,000-50,000 emergency department visits annually result from CO poisoning.

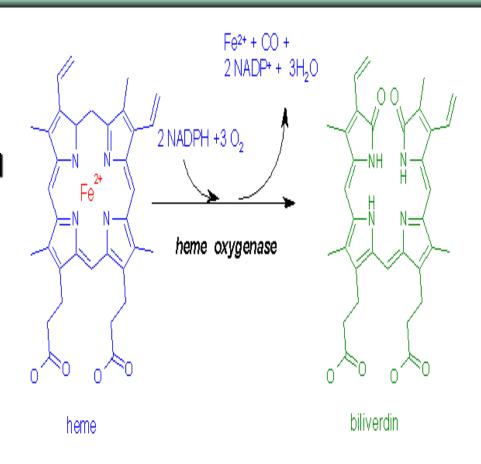
Tremendous amounts of carbon monoxide are released into atmosphere as a result of human activities. The combustion petroleum products remains by far the largest source of CO and the amount generated from petroleum product has risen.

- Endogenous
- Exogenous
- Methylene chloride



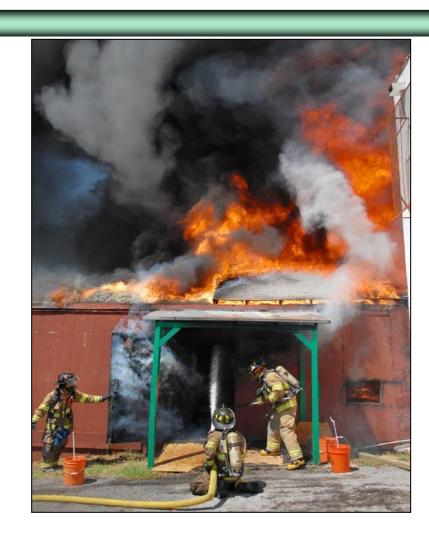
Endogenous:

- Normal heme catabolism:
 - Only biochemical reaction in the body is known to produce CO.
- Levels increase in:
 - Hemolytic anemia
 - Sepsis



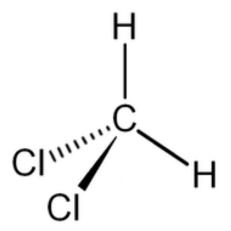
Exogenous:

- House fires.
- Gas-powered electrical generators.
- Automobile exhaust.
- Propane-powered vehicles.
- Heaters.
- Camp stoves.
- Boat exhaust.
- Cigarette smoke.



Methylene chloride:

- Paint and adhesive remover.
- Converted to CO in the liver after inhalation.





Motor vehicles have accounted for about 55% to 60% of global man-made emissions of carbon monoxide. This amount however is reduced in the past decades by using of catalytic converters. Another 20% of man-made CO emissions come from stationary sources such as space and water heaters and furnaces and from industrial processes, coal mine explosions, and solid waste disposal procedure. Because most plastics contain carbon, CO is one of the primary gases generated by heating and burning plastics.

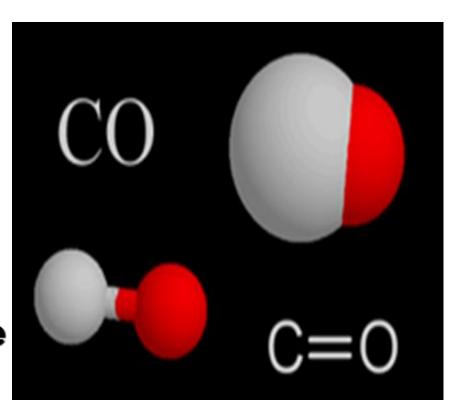
- Natural gas found with petroleum deposits has no CO, but carbon monoxide may be produced in processing natural gas (e.g. cracking).
- Additional sources of carbon monoxide include the manufacture of synthetic methanol and other organic compounds from CO, industrial and residential fires, charcoal burning, etc.



- **Epidemics** of carbon monoxide poisonings commonly occur **during winter months** and sources include misuse of non-electric heating or cooking devices.
- Tobacco smoke is also significant source of carbon monoxide, containing approximately 4% CO; smokers have been observed to have COHb levels typically in the 4 - 5% range.

Chemistry

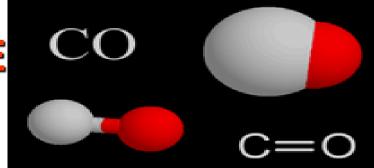
- Gas
- Odorless
- Colorless
- Tasteless
- Relative vapor density = 0.97
- Extremely stable
- Extremely flammable



- Half-life:
- Room air: 240-360 minutes
- O₂ (100%): 80 minutes
- Hyperbaric O₂: 22 min



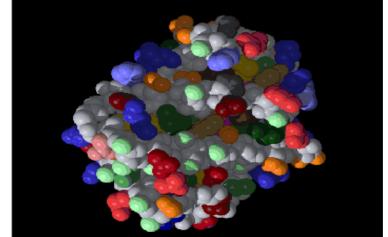
CARBON MONOXIDE Pathogenesis

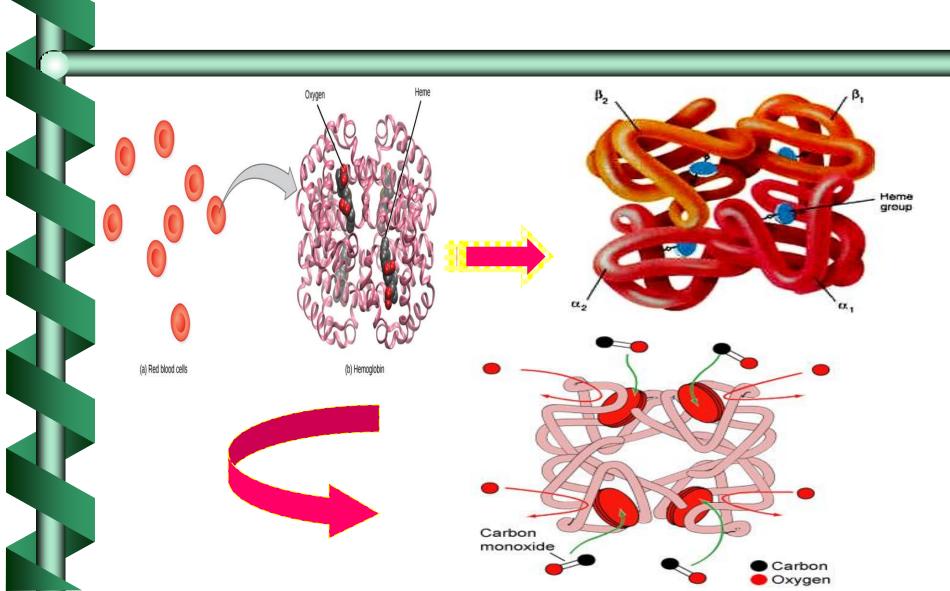


Carbon monoxide is non-irritating gas, resulting from incomplete burning of organic substances, as gasoline, coal products, woods, tobacco building materials.

CO is readily absorbed across the respiratory tract and binds to hemoglobin, forming carboxyhaemoglobin. The affinity of hemoglobin for carbon monoxide is some 250 times greater

than its affinity to oxygen.



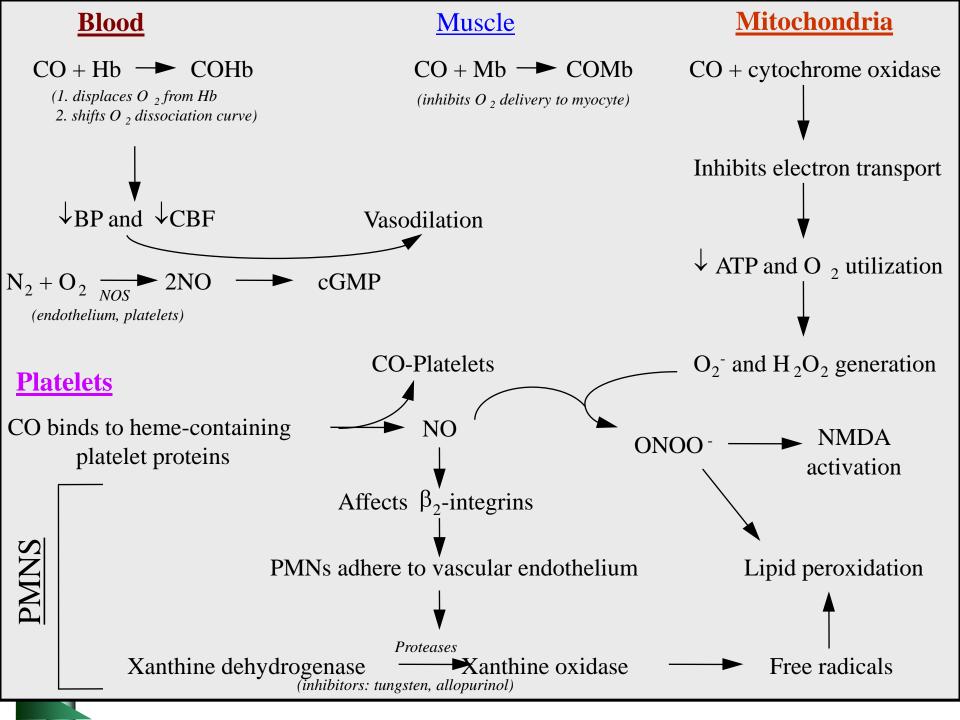


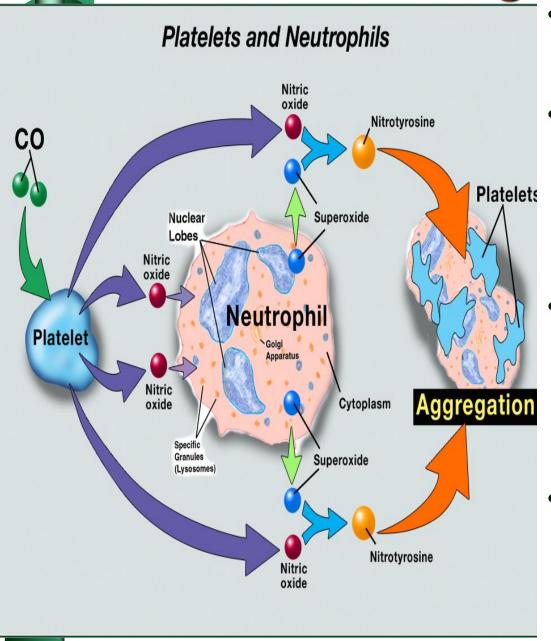
- The forming of the carboxyhaemoglobin reduces the capacity of blood to transport oxygen (OCC – oxygen-carrying capacity).
- Carboxyhaemoglobin is fully dissociable and once exposure has been terminated the pigment will revert to oxyhaemoglobin. Liberated carbon monoxide is eliminated via the lung.

- CO also binds to other ironcontaining proteins (Fe 2+):
 - Myoglobin (COMyoglobin) muscle cells
 - Cytochrome Oxidase
 - Neuroglobin
- Binding to myoglobin reduces O₂ available in the heart:
 - Ischemia
 - Dysrhythmias
 - Cardiac dysfunction

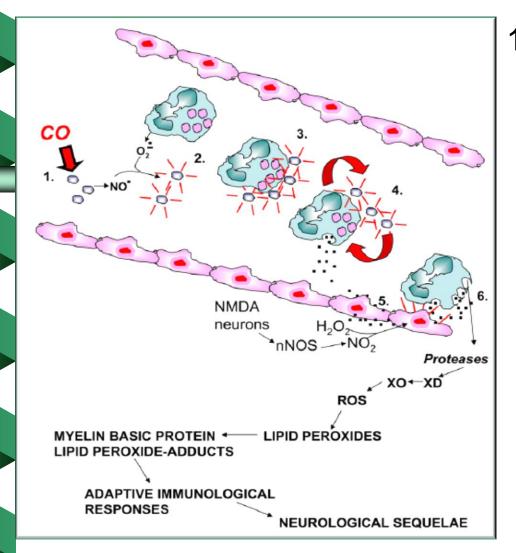
Other mechanisms:

- the binding of carbon monoxide with cytochrome oxydase
- cytochrome P-450





- CO binds to platelet hemoproteins and increases NO efflux.
- Platelet-derived NO reacts with neutrophilderived superoxide which activates platelets and causes plateletneutrophil aggregates.
- Reactive products and adhesion molecules promote firm aggregation and stimulate degranulation of neutrophils.
- Endothelial cells acitaved by myeloperoxidase facilitating firm neutrophil adhesion and further degranulation.



1.ROS initiate lipid peroxidation and adducts interact with brain myelin basic protein. The altered myelin basic protein triggers an adaptive immunologic response that causes neurologic dysfunction.

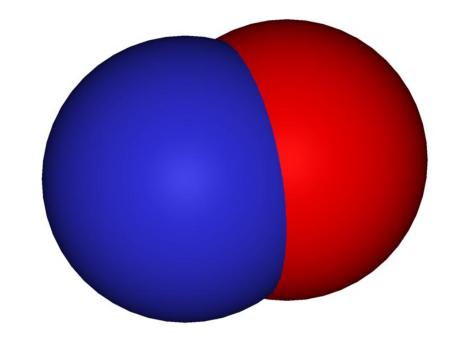
Source: Thom SR, Bhopale VM, Han S-T, Clark JM, Hardy KR. "Intravascular Neutrophil Activation Due to Carbon Monoxide Poisoning." *Am J Respir Crit Care Med.* 2006;174:1239-1248

Effects:

- CO exposure can cause:
 - Increased hydroxyl radicals noted during both the hypoxic and reoxygenation stage
 - CO causes hypoxia due to:
 - The direct effect on hemoglobin
 - Impaired perfusion from cardiac dysfunction.
 - CO impairs mitochondrial electron transport because CO binds to CcOX (at higher COHb levels).
 - Impairs brain ATP synthesis.

Nitric oxide (NO):

- Highly-reactive gas that participates in numerous biochemical reactions.
- Oxygen free-radical
- Levels increased with CO exposure.



Nitric Oxide (NO):

- Causes cerebral vasodilation:
 - Syncope (short loss of conciousness)
 - Headache
- May lead to oxidative damage to the brain:
- Probable cause of delayed neurologic syndrome (DNS).
- Associated with reperfusion injury.

- CO and NO are known second messengers
- CO, NO bind to heme and inhibit CcOX.
- NO targets intracellular heme.
- NO impairs heme synthesis and enhances heme destruction by increasing heme oxygenase activity.

- CO exposure can cause:
 - Increased NO levels
 - Increased superoxide levels
 - These can combine to form the highly toxic peroxynitrite.
 - Effect of free radicals is primarily on the vasculature.
 - May cause hemorrhagic necrosis.

Source: Ischiropoulos H, et al. "Nitric oxide production and perivascular tyrosine nitration in brain after carbon monoxide poisoning in the rat." *J Clin Invest.* 1996;97:2260-2267

Pathogenesis: CO Binding

Functional

Anemia

- Myocardium
 - CV impairment → Hypotension
- Hemoglobin
 - Decreased OCC(oxygen-carrying capacity) →
- Platelets and PMN
 - Nitric oxide → Hypotension
 - Free radicals
 → Lipid Peroxidation
- Mitochondria
 - Cytochrome oxidase → Lipid Peroxidation
 - Impaired e transport → Functional Hypoxia

Pathogenesis Summary

Limits O₂ transport:

 CO more readily binds to Hb forming COHb.

Inhibits O₂ transfer:

 CO changes structure of Hb (conformational changes) causing premature release of O₂ into the tissues.

Tissue inflammation:

Poor perfusion initiates an inflammatory response.

Pathophysiology Summary

Poor cardiac function:

- ↓ O₂ delivery can cause dysrhythmias and myocardial dysfunction.
- Long-term cardiac damage reported after single CO exposure.
- Increased activation of nitric oxide (NO):
 - Peripheral vasodilation.
 - Inflammatory response.

Pathogenesis Summary

Vasodilation:

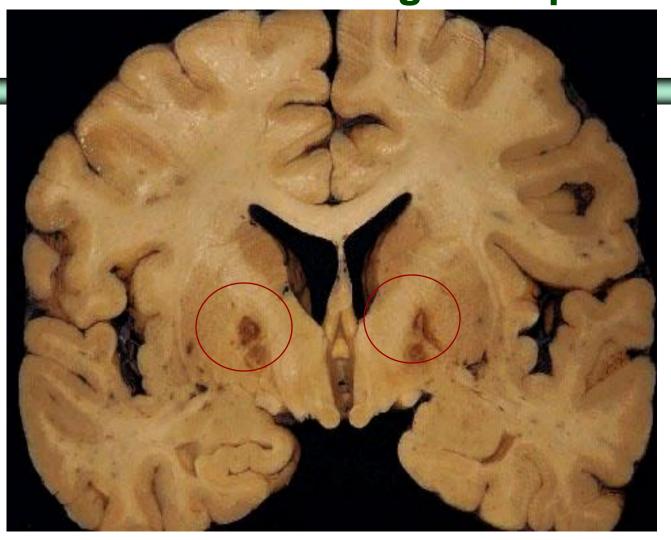
- Results from NO increase.
- Cerebral vasodilation and systemic hypotension cause reduced cerebral blood flow.
- NO is largely converted to methemoglobin.
- Free radical formation:
 - NO accelerates free radical formation.
 - Endothelial and oxidative brain damage.

Pathology

In patient who die early following carbon monoxide poisoning the brain is edematous, and there are diffuse petechias and hemorrhages. If the victim survives initially but dies within a few weeks, findings typical of ischemic anoxia are prominent.

- Manifests with cerebral edema, white matter petechial hemorrhages
- Congestion, hemorrhages and necrosis of globus pallidus

CARBON MONOXIDE TOXICITY: bilateral necrosis of globus pallidus



Acute effects

Initial symptoms are very nonspecific and may be mistaken for flu like symptoms.

In order of increasing severity, the most commonly reported signs and symptoms are:

- roaring in the ears
- weakness
- dizziness
- darkened vision
- frontal and temporal headache

- sleepiness
- adinamia
- collapse
- increased pulse and respiration
- unconsciousness
- involuntary evacuation
- muscle contractions
- coma
- convulsions
- cardiorespiratory depression
- death

Acute effects

 Pulmonary edema is commonly found in victims who die as a result of acute exposure. Aspiration pneumonia also is often seen, the result of vomiting. Ischemia or infarction of brain and heart can occur.

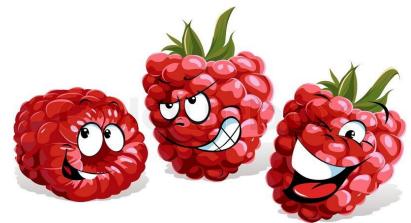
Acute effects

- The early features of CO poisoning are headache, dizziness, nausea and vomiting. Coma is accompanied by hyperventilation hypotension, increased muscle tone, hyperreflexia, clonus and shivering. The skin may show the cherry pink colour of carboxyhaemoglobin during life. Skin blistering may occur if the patient has been lying for some time. Renal failure is very rare. Retinal haemorrhages, hypoxic cerebral oedema may be present.
- Variety of neurological deficits such as parkinsonism, hemiparesis have been reported.
 The later correlate with lesions in the globus pallidus and putamen.

BLOOD







COHb levels (%)	Symptoms	
1-2	Normal	
5-10	Smokers	
10-20	"Flu-like" symptoms	
30-40	Fatigue; Severe headache	
40-50	Confusion; loss of consciousness	
60-70	Coma Seizures	CV collapse Death
>70	Rapidly fatal	

Summary

Impact of CO on major body systems:

Cardiac:

- Decreased myocardial function:
 - Hypotension with tachycardia.
 - Chest pain.
 - Dysrhythmias.
 - Myocardial ischemia.
 - Most CO deaths are from ventricular fibrillation.
- Long-term effects:
 - Increased risk of premature cardiac death.

Summary Impact of CO on major body systems:

- Metabolic:
- ◆ Respiratory alkalosis (from hyperventilation).
 Hyperventilation leads to low [CO2] in plasma and cerebral vasoconstriction, the oxygen supply to the brain is decreased.
- ♦ Metabolic acidosis with severe exposures (occurs secondary to lactic acidosis from ischemia).
- Respiratory:
 - Pulmonary edema (10-30%)
 - Direct effect on alveolar membrane.
 - Left-ventricular failure.
 - Aspiration.
 - Neurogenic pulmonary edema.
- Multiple Organ Dysfunction Syndrome (MODS):
 - · Occurs at high-levels of exposure.

Chronic effects

Some researchers believe that chronic effects actually are the result of a slow accumulation of daily damage that results in pathological change.

Symptoms of chronic CO effects that are most commonly seen include:

- headache
- irritability
- insomnia
- personality disturbances
- disturbance in ability to drive a vehicle

Delayed Neurologic Syndrome

Signs and Symptoms:

- Memory loss
- Confusion
- Ataxia
- Seizures
- Urinary incontinence
- Fecal incontinence
- Emotional lability

Signs and Symptoms:

- Disorientation
- Hallucinations
- Parkinsonism
- Cortical blindness
- Psychosis
- Gait disturbances
- Other motor disturbances

Chronic effects

- These manifestations are related to the effects of hypoxia on the nervous system. Basal ganglia lesions have been found in experimental animals and in autopsies of exposed humans.
- Exposure to low levels of carbon monoxide can provoke transient ischemic attacks, strokes and myocardial infarctions in workers with underlying cardiovascular or cerebrovascular diseases.

Diagnosis

 The diagnosis of carbon monoxide poisoning is most reliable confirmed by measuring the percentage of COHb in the blood. Typically this is measured in arterial blood but it is possible also in venous samples. A less accurate measure is the testing of expired air (for carbon monoxide)

Diagnosis

- Signs and symptoms closely resemble those of other diseases.
- Often misdiagnosed as:
 - Viral illness (e.g., the "flu")
 - Acute coronary syndrome
 - Migraine
- Estimated that misdiagnosis may occur in up to 30-50% of CO-exposed patients presenting to the ED.

Source: Raub JA, Mathieu-Holt M, Hampson NB, Thom SR. Carbon Monoxide Poisoning: A Public Health Perspective. *Toxicology* 200;145:1-14

Treatment

Treatment is based on the severity of symptoms. Individuals affected by carbon monoxide should be removed from exposure and moved to fresh air at once. If breathing has stopped, artificial respiration should be performed. If indicated, CPR should be completed.

Treatment generally indicated with SpCO > 10-12%. (pulse CO-oximetry is a continuous and noninvasive method of measuring the levels of CO concentration (SpCO) in the blood.

Be prepared to treat complications (i.e., seizures, dysrhythmias, cardiac ischemia).





TREATMENT

• The specific antagonist to carbon monoxide is the oxygen. Medical treatment consist of first supplying 100% supplemental oxygen via face mask the minute the diagnosis is suggested. Increasing the arterial oxygen content both counteracts the primary effect of the carbon monoxide and speeds its elimination. Immediate treatment with hyperbaric oxygen may be indicated.



Hyperbaric oxygen HBO

- It is involved 2 hours or longer at pressure of 2 3 ATA.
 It has been advised for patients who have:
- been conscious at any stage since exposure;
- COHb concentrations exceeding 40% at any time;
- neurologic or psychiatric features.
- * HBO produces a more rapid reduction in COHb levels. The half-life of COHb is 4-5 hours in a person at rest breathing room air, 80 minutes by administration of 100% oxygen at sea level and to 22-23 minutes by treatment with hyperbaric oxygen at 3 atmospheres absolute (ATA). At 3 ATA the oxygen dissolved in blood approaches 6 vol percent; this is adequate to supply basal oxygen requirements to the body with normal cardiac output in the absence of functional hemoglobin;

TREATMENT

- HBO induces cerebral vasoconstriction, which may reduce intracranial pressure and cerebral edema;
- HBO results in more rapid dissociation of CO from respiratory cytochromes;
- HBO may antagonize the oxidative injury that occurs after CO poisoning. HBO prevents the conversion of xanthine dehydrogenase to xanthine oxidase, a leucocyte-mediated reaction. This effect has been postulated to occur due to diminished B2 integrinmediated leukocyte adherence.

Hyperbaric chamber



TREATMENT

- In patient with altered level of consciousness naloxone (narcan) is administered.
- Exchange transfusion of blood has used for the moribund victims.
- If the patient has acidosis, an infusion of sodium bicarbonate is necessary until the pH of blood returned to 7,2.
- Giving i.v. of fluids to increase the circulatory blood volume and to correct hypotension.
- In case of convulsions is necessary diazepam (Valium).
- For preventing of ventricular dysrhythmias should be administered lidocaine.
- Mannitol and dexamethasone should be given if cerebral oedema is present.